

KOVALEV, Il'ya Denisovich; KORYAKIN, Yu.I., nauchnyy red.; NIKITINA,
T.K., red.; POPOVA, S.M., tekhn. red.

[Small-scale nuclear engineering in the United States] Malaia
atomnaia energetika v SShA. Möskva, dosatomizdat, 1963. 29 p.

(United States--Atomic power plants)

SINEV, N.M. (Moskva); KUVALEV, I.D. (Moskva)

Atomic electric power plant TES-3. Priroda 54 no.2:114-117 F 165.

(MIRA 18:10)

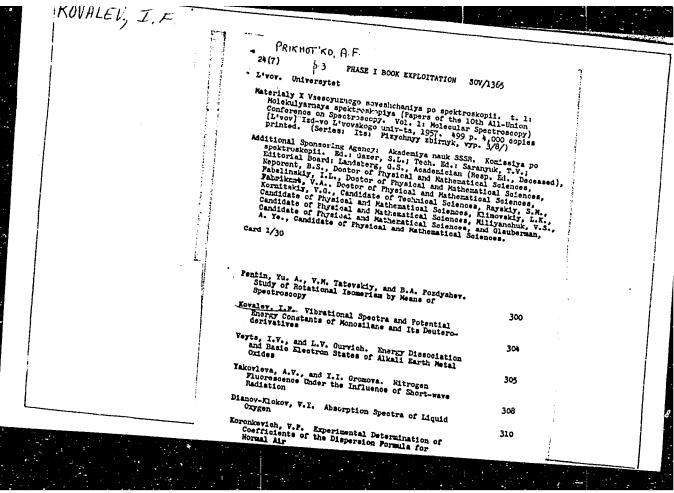
# WOVALEV, I.F. Spade for wrapping paper around a soil sample. Meteor.i gidrol. (MIRA 9:12) (Soils-Analysis)

APPROVED FOR RELEASE: 06/14/2000 CIA-RDP86-00513R000825610001-8"

KOVALEV, I. F.

Kovalev, I. F. — "Potential Functions of Methane and Ethane." Min Higher Education USSR, Moscow Engineering-Physical Inst. Moscow, 1955 (Dissertation for the Degree of Candidate in Physicomathematical Sciences)

SO: Knizhaya Letopis', No 24, 11 June 1955, Moscow, Pages 91-104



AUTHOR: Kovalev, I. F.

TITLE: Vibrational spectra and potential constants of monosilane and its deuterium derivatives. (Kolebatel'nye spektry i deyteroproizvodnykh).

PERIODICAL: "Optika i Spektroskopiya" (Optics and Spectroscopy), 1957, Vol.2, No.3, pp.310-316 (U.S.S.R.)

ABSTRACT: This theoretical paper presents calculations based on experimental values of the vibration spectra of SiH4, SiH2D2, Sind sind By the method of least squares the potential energy constants (the force and induction coefficients) were found for SiH4. Also the first derivatives of frequencies with respect to the force coefficients and the frequencies and forms of the normal Vibrations of monosilane and all the deuterated monosilanes were calculated. The silane molecule SiH4 is threedimensional and has T symmetry. The atomic mass of Si was taken to be 28.09 and the Si-H bond length 1.48 A; all angles were taken to be tetrahedral and the isotopic replacement was assumed not to affect the bond lengths or angles. The experimental data were taken from C. H. Tindal, Card 1/3 J. W. Straley, H. H. Nielsen (Phys. Rev., 62, 151, 1942) and J. Hawkins Meal, M. Kent Wilson (J. Chem. Phys., 24, 385, 1956)

51-6-2/25

## KOVALEV, I.F.

AUTHOR:

Kovalev, I. F.

TITIE:

Vibrational Spectra and Potential Energy Constants of Disilane and Hexadeuterodisilane. (Kulebatel'nyye spektry i postoyannyye potentsial'noy energii disilana

i geksadeyterodisilana.)

PERIODICAL: Optika i Spektroskopiya, 1957, Vol. III, Nr. 6,

pp. 552-559. (USSR).

Calculations of vibrational spectra of  $Si_2H_6$  and  $Si_2D_6$ ABSTRACT:

are carried out. Potential energy constants (force constants and induction coefficients) and normal vibration forms are calculated and discussed. ll fundamental

frequencies measured in Raman and infrared spectra of Si<sub>2</sub>H<sub>6</sub> (Refs.1,4,5) were used to calculate force constants

listed in Table 1. The reported SipD frequencies

(Ref.5) were used to check the calculation. Interpretation

of frequencies (including form of normal vibrations) is shown in Table 2. Correctness of this interpretation was checked by application of the product rule (Table 3).

Card 1/3

Card 2/3

APPROVED FOR RELEASE: 06/14/2000 CIA-RDP86-00513R60Q825610001-8 Vibrational Spectra and Potential Energy Constants of Disilane and Hexadeuterodisilane.

> The problem was dealt with using the method of Vol'kenshteyn et al. (Ref.6) and Mayants (Ref.7), using an ethane-like model D<sub>3d</sub> for the disilane molecule. The equilibrium lengths of bonds were taken to be 23.2 and 1.48 % for Si-Si and Si-H respectively. It was assumed that all angles were Tables 4 and 5 give the symmetrized tetrahedral. matrices of kinematic and force constants. The force constants were calculated by repeated approximation. The initial values were determined by the method of Ref.11. The force constants of disilane (in 10 cm -2) are given in a table at the top of p.556. These constants were used to find frequencies and forms of normal vibrations of Si<sub>2</sub>H<sub>6</sub> and Si<sub>2</sub>D<sub>6</sub> (Table 2). Table normal vibrations of Si<sub>2</sub>H<sub>6</sub> and Si<sub>2</sub>D<sub>6</sub> (Table 2). 6 gives the induction coefficients for Si2H6 and CH4, The author then compares the results  $C_2H_6$ ,  $SiH_4$ . obtained for disilane with those for ethane, and interprets

GOVRIEV, I.F.

USSR/Physical Chemistry - Molecule, Chemical Bond.

B-4

Abs Jour: Referat. Zhurnal Khimiya, No 3, 1958, 6925.

Author : I.F. Kovalev.

Inst : Physical Institute of Academy of Sciences of USSR.

Title : Computation and Interpretation of Vibration Spectra of

Methane and Its Deuterosubstitutes.

Orig Pub: Zh. fiz. khimii, 1957, 31, vyp. 2, 362-371.

Abstract: Bibliographical and experimental data concerning Raman spectra and infrared absorption spectra of CoHo, CoHo, CoHo, and CoDo served as the foundation for the computation. Referring the frequencies of these molecules to the concepts of symmetry groups, the sensitivity of these frequencies to various parameters were computed, and the influence of isotope substitutions on the frequencies was investigated by the method of L.S. Mayants (Tr. Fiz. in-ta AN SSSR, 1950, 5, 63). At the formation

of secular equations, the inharmoniousness was taken into consider-

Card : 1/3

-15-

Sov/51-4-4-6/24

AUTHORS: Babushkin, A.A., Kovalev, I.F. and Yemel'yanova, V.M.

TITLE: Investigation of the Vibrational Spectra of Molecular Compounds of Boron Trifluoride with Substances Containing

Nitrogen and Oxygen. I. F3B.NH3 and F3B.ND3

(Issledovaniye kolebatel'nykh spektrov molekulyarnykh soyedineniy trekhftoristogo hora s azot- i kislorod-soderzhashchimi veshchestvami. I. F<sub>3</sub>B.NH<sub>3</sub> i F<sub>3</sub>B.ND<sub>3</sub>)

PERIODICAL: Optika i Spektroskopiya, 1958, Vol IV, Nr 4, pp 468-473 (USSR).

ABSTRACT: Boron trifluoride was obtained by decomposition of  ${}^{C_6}{}^{H_5}{}^{N_2} \cdot {}^{EF_3}$ . Purity of boron trifluoride was checked spectrally and only  ${}^{SiF_4}$  in an amount smaller than 0.5% was found.

Molecular compounds of boron trifluoride with ammonia and deuteroammonia were obtained by condensation of ammonia (or deuteroammonia) on freezing by means of liquid nitrogen in an ether complex of boron trifluoride,  $(c_2H_5)_2O_{\circ}BF_3$ , in a

metal test tube. A white crystalline substance was obtained which was re-crystallised in water (or heavy water) or in acetone. In re-crystallisation of F<sub>3</sub>B.ND<sub>3</sub> from acetone, a Cardl/4

Sov/51-4-4-6/24 Investigation of the Vibrational Spectra of Molecular Compounds of Boron Trifluoride with Substances Containing Nitrogen and Oxygen. I. F<sub>3</sub>B.NH<sub>3</sub> and F<sub>3</sub>B.ND<sub>3</sub>

replacement of deuterium by hydrogen occurred and a mixture of compounds with different degrees of replacement of hydrogen, by deuterium was obtained. This mixture was denoted by the formula F3B.NHiDk, where i and k may have the values O, 1, 2, 3 and i + k = 3. The spectra were recorded using a spectrometer IKS-11 in the region from 2.5 to 15  $\mu$  . To avoid absorption by atmospheric water vapour and carbon dioxide, nitrogen was passed through the spectrometer. Samples were prepared by placing a layer of paste of the substance studied between two plates of rock-salt or by placing a dry layer of the substance between the same plates. Raman scattering spectrum of an aqueous solution of the molecular compound  $F_{\bar{z}}\bar{B}_*N\bar{H}_{\bar{z}}$  was recorded by means of a spectrograph ISP-51 with a photoelectric attachment UF-320. The infra-red absorption spectra of F3B.NH3 and F3B.ND3 are shown in Figure 1. Calculations of the force field and vibrational spectra were based on molecular models with  $C_{3v}$  symmetry for  $F_3B.NH_3$  and  $F_3B.ND_3$  (Figure 2) Card2/4

Sov/51-4-4-6/24 Investigation of the Vibrational Spectra of Molecular Compounds of Boron Trifluoride with Substances Containing Nitrogen and Oxygen. I. F3B.NH3 and F3B.ND3

and  $C_s$  symmetry for  $F_3B.NH_2D$  and  $F_3B.NHD_2$ . the force constants, the authors used their own experimental results on the Raman and infra-red spectra of  $F_3B.\overline{N}H_3$  and (see table on p 471). The observed frequencies for  $F_zB.ND_z$ the mixture denoted by F3B.NHiDk were used to check the The force field for  $F_3B.NH_3$  was calculated calculations. by the method of Vol'kenshteyn, Yel'yashevich, Stepanov (Ref 18) and Mayants (Ref 13) using "spectroscopic masses" for hydrogen and deuterium. From 49 force constants, which determine the potential function 18 were taken to be equal to zero. The calculated force constants are given at he top of p 473. They were calculated using the BESM computer of the Ac.Sc. USSR. The table on p 471 shows that there is good agreement between the observed frequencies and those obtained by calculation using the force constants, Card3/4 authors also calculated the coefficients of induction which

Sov/51-4-4-6/24 Investigation of the Vibrational Spectra of Molecular Compounds of Boron Trifluoride with Substances Containing Nitrogen and Oxygen. I.  $F_3B.NH_3$  and  $F_3B.ND_3$ 

are given in the middle of p 473. The authors thank A.I. Shatenshteyn for supply of deuterated ammonia and K.I. Podlovchenko for help in carrying out the calculations on the computer. There are 2 figures, 1 table and 19 references, 6 of which are Soviet, 8 in English, 2 German, 1 French, 1 translation of Western work into Russian and one other.

ASSOCIATION: Institut fizicheskoy khimi AN SSSR (Institute of

Physical Chemistry, Ac.Sc. USSR), Saratovskiy pedagogicheskiy institut (Saratov Pedagogical

Institute) and Moskovskiy gosudarstvennyy universitet

(Moscow State University)

SUBMITTED:

June 14, 1957

4/4 Card 4/4

1. Boron florides--Spectra

AUTHOR:

Kovalev, I.F.

51-4-5-3/29

TITLE:

Vibration: Spectra and Potential Energy Constants of Methylsilane and Methyltrideuterosilane (Kolebatoliniye spektry i postoyannye potentsial'noy energii metilsilana i metiltrideyterosilana,

PERIODICAL: Optika i Spektroskopiya, 1958, Vol IV, Nr 5, pp 560-508 (USSR)

ABSTRACT:

The force field and vibrational spectra of  $\mathrm{CH_3SiH_5}$  and  $\mathrm{CH_3SiD_3}$ were calculated using the method of Vol'kenshteyn et al. (Ref. 17) and Mayants (Ref. 20), using "spectroscopic masses" for hydrogen and deuterium. The calculations were made using the following geometrical parameters for the methylsilane molecule (Ref. 2);; r(Si-C) = 1.857 &, r(C-H) = 1.09 &, r(Si-H) = 1.48 Å, with all angles totrahedral. Values of the kinematic and dynamic coefficients in symmetry coordinates are given in Table 1. To find the force-field coefficients the author used results of the calculations for ethane (Ref. 17), disilane (Ref. 14), as well as experimental frequencies (Table 2). Force constants and induction coefficients were calculated for CH\_SiH3. Distribution

Card 1/2

Vibrational Spectra and Potential Energy Constants of Methylsilane and Methyltrideuterosilane

of forces inside the CH<sub>3</sub>SiH<sub>3</sub> molecule was found to differ considerably from the force distributions in ethane and disilane. Both frequencies and forms of the normal vibrations of methylsilane were calculated. They are given in Table 2, which shows that good agreement was obtained with experimental values (the mean absolute error does not exceed 5 cm<sup>-1</sup>). The vibrational frequencies for methyltrideuterosilane (Table 2, last column) were also calculated. A full interpretation of the infrared vibrational spectrum of methylsilane is given. The author thanks M. M. Sushchinsky for his interest. There are 1 figure, 3 tables and 21 references, of which 11 are American, 8 Soviet, 1 German and 1 Grech.

ASSOCIATION: Saratovskiy pedagogicheskiy institut (Saratov Pedagogical Institute)

SUBMITTED: July 4, 1957

Card 2/2

1. methylsilane - Vibration spectra 2. methylsilane - energy 3. methyltrideuterosilane - Vibration spectra

4. Methyltrideuterosilane - Energy

AUTHORS:

Babushkin, A. A., Kovalev, I. F., Yemel'yanova, V. M.

SOV/48-22-9-33/40

TITLE:

Spectroscopic Investigations of the Structure of Some Complex Compounds (Spektroskopicheskiye issledovaniya stroyeniya nekotorykh kompleksnykh soyedineniy)!. Molecular Compounds  $F_3B.NH_3$  and  $F_3B.ND_3$  (1. Molekulyarnyye soyedineniya  $F_3B.NH_3 i F_3^2B.ND_3^2$ )

PERIODICAL:

Izvestiya Akademii nauk SSSR. Seriya fizicheskaya, 1958, Vol 22, Nr 9, pp 1131 - 1131 (USSR)

ABSTRACT:

This is a condensation of he paper which was published under the above subtitle Nr 1 in the "Izvestiya Akademii nauk SSSR" by A.A.Babushkin. The spectra of infrared absorption and of combination dispersion of the compounds in question were recorded in the laboratory of the Institut fizicheskoy khimii Akademii nauk SSSR (Institute of Physical Chemistry AS USSR). The field of force and the vibration spectra were computed by I.F.Kovalev. The computations were based upon the model  $c_{5y}$  for  $F_3B.NH_3$  and

Card 1/2

upon the model C<sub>S</sub> for F<sub>3</sub>B.NH<sub>2</sub>D and F<sub>3</sub>B.NHD<sub>2</sub>. The spectrum

<sup>Са</sup> АРР ROVED FOR RELEASE: 06/14/2000 CIA-RDP86-00513R000825610001-8

24(7), 24(4)

AUTHOR:

Kovalev, I.P.

-0V/51-6-5-6/3-

TITLE:

Calculation and interpretation of the Vibrational Spectra of Tebranethylsilane and Tetramethylsilane- $d_{12}$  (Raschet i interpretative colobatel nyth spektrov tetrametilaisana i tetrametilaisana- $d_{12}$ )

MERICORDAL: Optika i Septemberopiya, 1950, Vol 6, Nr 5, pp 534-599 (USER)

.B3Th.OT:

The potential energy constants and frequencies of normal vibrations of methyloilanes are of considerable interest in theoretical and experimental chemistry. Theoretical calculations of the spectral properties of SigN<sub>d</sub> (sef 3) and ONGOING (Ref 4) were already reported. The present paper deals with tetramethyloilane Si(CHg)4 which has form aid bonds. The author used experimental data on frequencies, degrees of depositivation and intensities of the Raman and infrared spectra of depositivation and intensities of the Raman and infrared spectra of tetramethyloidane reported by various workers (Refs 8-17). The secondary constants (force constants and "industion coefficients", and for a of account vibrations of tetramethyloidane were coloulated asing the between all vibrations of tetramethyloidane were coloulated asing the between the lateral process of tetramethyloidane were coloulated asing the between the lateral process of tetramethyloidane were coloulated asing the lateral (loss 8). The tramethyloid of the vibrational spectra of

Ja d 1/2

SOV/51-6-5-6/34

Calculation and Interpretation of the Vibrational Spectra of retramethylsilane and Tetramethylsilane- $\mathbf{d}_{12}$ 

Si(CH<sub>3</sub>)<sub>4</sub> and Si(CD<sub>3</sub>)<sub>4</sub> were also calculated and interpreted. The calculated data are listed in Tables 1 and 2 (the latter gives the interpretation of the combination frequencies and harmonics in the infrared spectrum of tetramethylsilane). The calculated frequencies agreed well with the experimental ones, showing that the potential energy constants used describe accurately the force field of the tetramethylsilane molecule. Acknowledgment is made to M.M. Sushchinskiy for his advice. There are 2 tables and 18 references, 7 of which are Soviet, 10 English and 1 German.

SUBMITTED: June 7, 1958

Card 2/2

5/051/60/008/03/007/038 E201/E191

AUTHOR: Kovalev, I.F.

Vibrational Spectral and Potential Functions of TITLE:

Methylsilanes CH3SiH3, (CH3)2SiH2 and (CH3)3SiH

PERIODICAL: Optika i spektroskopiya, 1960, Vol 8, Nr 3,

pp 315-323 (USSR)

ABSTRACT: This paper is one of a series dealing with the force

fields and vibrational spectra of the compounds  $(CH_3)$   $nSiH_{4-n}$ , where n = 1, 2, 3 and 4. Earlier papers

dealt with SiH4 (Ref 1) and (CH3)4Si (Ref 2); the first calculations for CH3SiH3 were also reported (Ref 3). The present paper describes calculations and reports more accurate values of the fundamental frequencies (Tables 2 to 5) and the force constants (Table 1) of methylsilane

CH3SiH3, dimethylsilane (CH3)2SiH2 and trimethylsilane The force fields, frequencies and forms of (CH<sub>3</sub>)3SiH. normal vibrations were calculated using the method of

Vol'kenshteyn, Yel'yashevich, Stepanov (Ref 4) and Mayants (Ref 5). The calculations were carried out simultaneously for the whole series of methylsilanes

Card 1/2

S/051/60/008/03/007/038 E201/E191

Vibrational Spectra and Potential Functions of Methylsilanes CH3SiH3 (CH3)2SiH2 and (CH3)3SiH

(CH3) nSiH<sub>4</sub>-n using the results obtained earlier for ethane and disilane (Ref 3). The "spectroscopic mass" of hydrogen was used. Interpretation and comparison of the published experimental results with the calculated vibrational spectra of CH3SiH3 (Table 2), (CH3) 2SiH2 (Tables 3 and 4), and of (CH3) 3SiH (Tables 5 and 6) show good agreement between experiment and calculation. Acknowledgements are made to M.M. Sushchinskiy for his advice, and to D. Marais (Union of South Africa), M. Randič (Yugoslavia) and N. Sheppard (England) for supplying their experimental results on the spectra of CH3SiH3 before publication.

Card 2/2 publication.
There are 2 figures, 6 tables and 12 references, of which 7 are Soviet, 3 English, 1 Scandinavian and 1 private communication.

SUBMITTED: July 11, 1959

### KOVALEV, I.F.

Potential functions of molecules of the homologous series (CH<sub>3</sub>) $nSiH_{4-n}$  (n= 1 - 4). Dokl.AN SSSR 134 no.3:559-562 S '60. (MIRA 13:9)

1. Saratovskiy gosudarstvennyy pedagogicheskiy institut. Predstavleno akad. I.V. Obreimovym.

(Silicon organic compounds) (Chemical structure)

## Vibration spectrum and force field of the trimethylchlorosilane molecule. Opt. i spektr. 10 no.6:707-712 Je '61. (MIRA 14:8) (Silane—Spectra)

20633

\$/020/61/136/006/009/024

5.3100

2209, 1153, 1372

B104/B204

AUTHOR:

Kovalev, I. F.

TITLE:

Potential functions of the molecular series  $(CH_3)_n SiCl_{4-n}$ 

(n = 1-4)

PERIODICAL:

Doklady Akademii nauk SSSR, v. 136, no. 6, 1961, 1313-1316

TEXT: The author calculated the coefficients of force and effect between the atoms of the molecules, the frequency and shape of normal escillations, and interpreted the experimentally determined spectra. In the present paper, the results of calculating the coefficients of the above-mentioned effect are given. The following parameters (given in A) were taken from non-Soviet papers:

OngSiCl <sub>3</sub>	r(C-H) 1.093	r(Si-C) 1.876	r(S1-Ci 2.021
(OH <sub>3</sub> ) <sub>2</sub> sicl <sub>2</sub>	1.093	1.83	1.99
(CH <sub>3</sub> ) <sub>3</sub> sicl	1.093	1.89	2.09

Card 1/8

20633

Potential functions of the ...

S/020/61/136/006/009/024 B104/B204

The denotation of the coordinates is given in Fig. 1; in Table 1 the coefficients of the above-mentioned effect are written down. When using these coefficients for calculating the oscillations, results are obtained, which show good agreement with experimental observations. The following conclusions are drawn: 1) When changing from tetramethylsilane to methylchlorosilane, the coefficients  $(Q_1,Q_4)$  decrease

systematically, which is related with the extension of the distance Si-C. If one of the methyl groups is replaced by a chlorine atom, the jump is particularly large. A spectrum analysis shows an increase of the frequency of natural stretching vibrations. This is connected with the intensification of the covalent character of the bonds. 2) When the chlorine atoms are surrounded by silicon, a slight but systematic decrease of the coefficients is observed, which is related with the change in the length of the bond C - H. 3) The reciprocal bond Si-Cl leads to an increase of its strength. 4) The strength of the inner and outer angles of the CH<sub>3</sub> group is somewhat diminished. There are

1 figure, 1 table, and 14 references: 6 Soviet-bloc and 8 non-Soviet-bloc.

Card 2/8

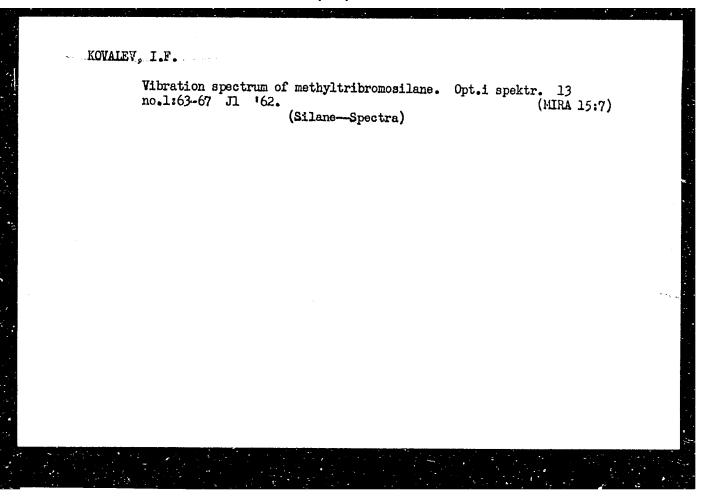
(Saratov State Pedagogical Institute) PRESENTED: September 14, 1960, by I. V. Obreimov, Academician	ASSOCIATION:	Saratovskiy gosudarstvennyy pedagogicheskiy institut	
SUBMITTED: September 10, 1960  Regend to Fig. 1: Equilibrium configuration of the (CH <sub>3</sub> ) <sub>3</sub> SiCl molecule.  H <sub>3</sub> Q <sub>3</sub> Q <sub>4</sub> H <sub>4</sub> Q <sub>4</sub> Q <sub>5</sub> H <sub>5</sub> H <sub>7</sub> H <sub>4</sub> H <sub>4</sub> H <sub>4</sub> H <sub>7</sub> H <sub>8</sub> Q <sub>4</sub> H <sub>7</sub> H <sub>8</sub> H <sub>9</sub> Q <sub>4</sub> H <sub>7</sub> H <sub>9</sub> Q <sub>4</sub> H <sub>7</sub> H <sub>9</sub> Q <sub>4</sub> H <sub>8</sub> H <sub>9</sub> Q <sub>4</sub> H <sub>9</sub> H <sub>1</sub> H <sub>2</sub> Red and H <sub>1</sub> H <sub>2</sub> H <sub>1</sub> H <sub>1</sub> H <sub>1</sub> H <sub>2</sub> H <sub>1</sub> H <sub>1</sub> H <sub>2</sub> H <sub>1</sub> H <sub>2</sub> H <sub>3</sub> H <sub>4</sub> H <sub>1</sub> H <sub>1</sub> H <sub>1</sub> H <sub>1</sub> H <sub>2</sub> H <sub>3</sub> H <sub>4</sub>			
Regend to Fig. 1: Equilibrium configuration of the (CH <sub>3</sub> ) <sub>3</sub> Sicl molecule.  H <sub>3</sub> Q <sub>1</sub> H <sub>3</sub> H <sub>4</sub> H <sub>4</sub> H <sub>4</sub> H <sub>4</sub> H <sub>4</sub> H <sub>5</sub> H <sub>7</sub> H <sub>8</sub> H <sub>9</sub> H <sub>1</sub> H <sub>2</sub> H <sub>3</sub> H <sub>4</sub> H <sub>5</sub> H <sub>7</sub> H <sub>8</sub> H <sub>9</sub> H <sub>1</sub> H <sub>2</sub> H <sub>3</sub> H <sub>4</sub> H <sub>5</sub> H <sub>7</sub> H <sub>8</sub> H <sub>9</sub> H <sub>1</sub> H <sub>2</sub> H <sub>3</sub> H <sub>4</sub> H <sub>1</sub> H <sub>1</sub> H <sub>1</sub> H <sub>1</sub> H <sub>2</sub> H <sub>3</sub> H <sub>4</sub> H <sub>4</sub> H <sub>5</sub> H <sub>7</sub> H <sub>8</sub> H <sub>8</sub> H <sub>9</sub> H <sub>9</sub> H <sub>1</sub> H <sub>2</sub> H <sub>1</sub> H <sub>2</sub> H <sub>1</sub> H <sub>2</sub> H <sub>3</sub> H <sub>4</sub> H	PRESENTED:	September 14, 1960, by I. V. Obreimov, Academician	
Equilibrium configura- sion of the $(CH_3)_3$ SiCl nolecule.  H <sub>3</sub> $Q_1$ $Q_2$ $Q_3$ $Q_4$ $Q_5$ $Q_5$ $Q_5$ $Q_7$ $Q_7$ $Q_7$ $Q_8$	SUBMITTED:	September 10, 1960	•
sion of the $(CH_3)_3$ SiCl molecule. $H_3 \qquad \qquad H_4 \qquad \qquad H_5 \qquad \qquad H_5 \qquad \qquad H_6 \qquad \qquad H_7 \qquad \qquad H_7 \qquad \qquad H_8 \qquad \qquad H_9 \qquad H_9 \qquad \qquad $			1
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	ion of the (C	$\begin{array}{c} \text{niigure-} \\ \text{H}_3)_3 \text{SiCl} \\ \end{array}$	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	notecule.		•
$\begin{array}{c} H_2 \xrightarrow{\alpha_{23}} G_2 \\ G_1 \\ G_2 \\ H_4 \end{array}$ ard 3/8	1. Sec. 1.	H <sub>3</sub> Q <sub>3</sub> $\stackrel{\text{H}_5}{\bigcirc}$ $\stackrel{\text{H}_9}{\bigcirc}$	
$\begin{array}{c} G_1 \\ G_7 \\ H_1 \end{array}$		~ \ E \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \ \	$\mathcal{N}$
ard 3/8	•	H <sub>2</sub> 23 + F3	V
ard 3/8		$H_2$ $C_1$ $H_6$ $C_2$ $H_5$ $H_7$	V
		$H_2$ $G_1$ $G_3$ $G_4$ $G_5$ $G_5$ $G_6$ $G_7$ $G_8$	V

## Royalev, I.F.

Potential functions of molecules of the homologous series (CH<sub>3</sub>)n SiBr<sub>4-n</sub>(n = 1 - 4). Dokl. AN SSSR 142 no.5:1069-1072 F 162. (MTRA 15:2)

1. Saratovskiy pedagogicheskiy institut. Predstavleno akademikom I.V.Obreimovym. (Silane)

# Calculation of intensities in the infrared spectra of molecules of the type XY3ZV3. Opt.i spektr. 12 no.5:5500-556 My '62. (MIRA 15:5) (Molecular spectra, (Silane)



# Calculation of normal vibrations of dimethyldibromsilane and trimethylbromsilane. Opt. i spektr. 13 no.3:335-340 S :62. (MIRA 15:9)

L 17791-63 EPB/EWP(1)/EPF(c)/EWT(b)/BDS ASD ACCESSION NR: AP3005839 8/0051/83/015/002/0186/018 AUTHOR: Kovalev, I.F. TITLE: Vibrational spectrum and force field of methyl trifluorosilane SOURCE: Optika i spektroskopiya, v.15, no.2, 1963, 186-189 TOPIC TAGS: force constant, vibrational spectrum, mothyl silane Investigation of the physical-chemical properties of simple organosili con compounds is of considerable interest from the standpoints of molecular spectroscopy and chemistry in genera; Earlier the author carried out theoretical studies of the force fields and vibrational parameters of methyl fluorosilanes (I.F. Kovalev, Optika i spektroskopiya, 8,315,1960 and Doklady AN SSSR,134, 559, 1960) and of their chlorine and bromine substituted analogs (Doklady AN SSSR, 136, 1313, 1961; Optika i spektro., 12, 11, 1962; Doklady AN SSSR, 142, 1069, 1962; Optika i spoktro.,13, 63, 1962). The present paper gives the results for methyl trifluoro silane  $ext{CH}_3 ext{SiP}_3$ , obtained as part of a comprehensive investigation of the  $ext{(CH}_3 ext{)}_n$ - $SiR_{4-n}$  (n = 1 to 4) homologous series. The parameters calculated were the force constants and the "influence" (coupling) coefficients. For  $CH_3SiF_3$  the author used the data of R.L.Collins and J.R.Nielsen (J.Chem.Phys., 23, 351, 1955) on the infra-

Card 1/2

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ed and Raman spectra of id of "Strela" computer	this compound. The secular solutions w . The force constants for the various b	ere solved with the	
alculated on the basis ata. The vibrational s ormation vibrations of	of the force constants, are tabulated to pectra are interpreted and the different	ic displacements, gether with other stretching and de-	
ne molecule are compared talogs. "The author is	with the respective parameters in its	thyl trafluorosil-	
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19 molecule are company	with the respective parameters in its of grateful to <u>I.V.Obreimov</u> for his interestry, art.has; 3 tables.	thyl trafluorosil-	
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### KOVALEV, I.F.

Examination of vibration spectra of dimethyldifluorosilane and trimethylfluorosilane molecules. Call Cz Chem 28 no.6: 1364-1373 Je '63.

1. Saratovskiy Gosudarstvennyy pedagogicheskiy Institut, Saratov, SSSR.

\$/020/63/148/003/018/037 B191 B102

AUTHOR:

Kovalev, I. F.

TITLE:

Card 1/2

Potential functions of the molecules of the homologuous

series  $(CH_3)_n SiF_{4-n}$  (n=1,..4)

PERIODICAL:

Akademiya nauk SSSR. Doklady, v. 148, no. 3, 1963,

569-572

TEXT: The methods proposed by M.V. Vol'kenshteyn et al. (Kolebaniya molekul - Molecular vibrations -, 1, M.1949) and L.S. Mayants (Teoriya i raschet kolebaniy molekul - Theory and calculation of molecule vibrations -, 1960) are used for calculating bond strengths and mutual effects of the molecule groups in  $(CH_3)_n SiF_{4-n}$  compounds. The forces due to vibrational changes in the bond lengths r are also determined. For  $CH_3SiF_3$ ,  $(CH_3)_2SiF_2$  and  $(CH_3)_3SiF_1(C-H) = 1.10, 1.09, 1.093; <math>r(Si-C) = 1.10$ 1.88, 1.89, 0.87; and r(Si-F) = 1.55, 1.56, 1.55; r is given in A for the equilibrium positions. The numerical results cover a table of two pages; they show how the forces and the mutual effects change from one

Potential functions of 'the ...

S/020/63/148/003/018/037 B191/B102

compound to the other. This change is governed by a certain law: If a methyl group is replaced by F atoms the Si-C bond strength increases by about 10%. Similar effects are also observed on Cl or Br substitution. If halide atoms are added to Si atoms an induction effect is observed, i.e. the molecular electron shell is displaced toward the halide. The Si-F bond strength increases with the number of F atoms at the Si. The C-H bond strength and the HCH angles remain almost equal for all molecules. The valency  $\nu$  (C-H) and inner stretching vibrations  $\delta$  (CH<sub>3</sub>) in methylhalide-

silanes are characteristic in frequency and shape. The changes in bond length due to these vibrations are calculated with an accuracy of 0.0001 Å. These changes are of the order of 0.001 to 0.01 Å and depend very little on n, except for  $\nu(\text{Si-Hal})$ , where they amount to 0.0263 for n=1, 0.0096 for n=2 and 0.0035 for n=3. There are 1 figure and 3 tables.

ASSOCIATION:

Saratovskiy gosudarstvennyy pedagogicheskiy institut

(Saratov State Pedagogical Institute)

PRESENTED:

August 1, 1962, by I.V. Obreimov, Academician

SUBMITTED:

August 1, 1962

Card 2/2

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ACCESSION NET AR5012230 UR/0058

UR/0058/65/000/003/D014/D014

SOURCE: Steff 21 Platte, 10s 3789

AUTHOR: [ Necessor, V. P.; Nevalev, 1:7: | Teauni, A. Na.; Malebnikova, V.B.; Ayasra, N.I.; Novallohuk, U. S.

Tiving: enlowing ion of vibrational apactra of simple molecules with appoint of sunharmonicity

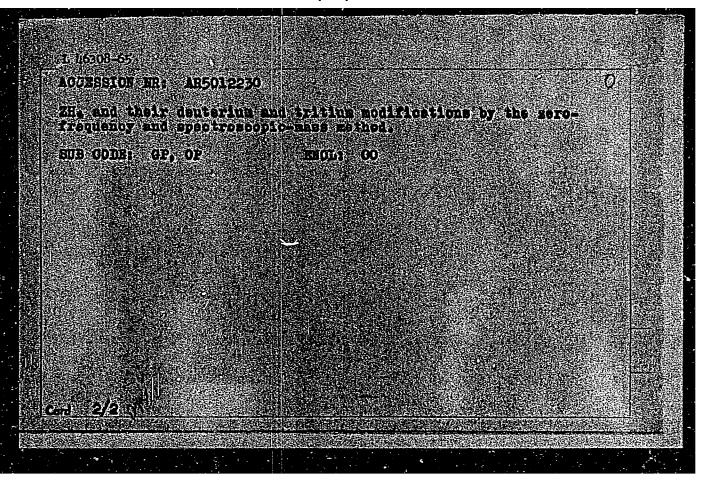
OLTED SOURCE: Tr. Komis, Do spektroskopii. All SSEE, vyp. 1, 1964, 170-176

TOPIC TAGS: wibrational spactrum, anharmonicity electro-optical parameter, force constant, intrared intensity

ICANSMITION: Methods of quantum mechanics are used to justify linear classical methods accounting for anharmonicity. A derivation is given for direct year and a formulas for the calculation of the sleepzo-optical parameters of molecules of the type XI<sub>2</sub>Z<sub>2</sub> and XI<sub>2</sub>Z. The force constants the vibrational traquencies the forms of oscillastions, the sientro-optical parameters, and the intensities are baloniated for the infrared spectra of hydrides of the type IH<sub>2</sub>. The

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ACCESSION NET AP4043879

**8/0139/64/000/004/0173/0179** 

AUTHOR: Rovalev, I. P.

TITUE: Calculation of <u>electro-optical</u> parameters and intensities in the infrared spectra of molecules of the type XY4, XY32, and XY22,

SOURCE: IVIE. Fisika, no. 4, 1964, 173-179

TOPIC TAGS: infrared spectrum, electro optical parameter, vibretion spectrum, tetrahedral molecule, crystal symmetry, molecular electronics

ABSTRACT: In view of their importance to the theory of the structure of matter, the author derives general formulas for the calculation of the electro-optical parameters and band intensities of infrared absorption vibration spectra for several simple molecules of the type  $XY_1Z_4$ ,  $\{n=2-4\}$ . Three cases are considered: 1) arbitrary angles between bonds; 2) tetrahedral angles; 3) isotopic series of

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ACCESSION NR: AP4043879	
molecules. For the case of tetrahedral matrices are expressed in terms of the subsplacements of the atoms in normal vib parameters, and the intensities are calculated tetrahedral molecules, relations are der dependence of the atomic displacements of the lengths of the bonds. These form future article to monosilans and its deur art. his: I figure and 5 formulas	ymmetry coordinates. The rations, the electro-optical ulated. In the case of ived showing the general names are
ASSOCIATION: Saratovskiy gospedirstitut Institute)	(Saratov State Pedagogical
SUBATTAD, 23Apr63	BNCL: 00
SUB CODE SS, OP WR REF SOV.	.004 OTHER: 000
Cord 2/2	

8/0139/64/000/005/0102/0105 ACCESSION NR: AP4047355 AUTHOR: Kovalev, I. F. TIVIE; Calculation of electro-optical parameters and intensities in the it parties of motion is of the type XY, XX,E, and XY, Z2. The ser es 6/20 \_\_\_ (n - 0-4) SOURCE: TVUZ: Fizika, no. 5, 1964, 102-105 TOPIC TAGE: electrosptical parameter, it intensity, ir spectrum, monosilane ABSTRACT: Using the connection between the natural coordinates Charges in the equilibrium bond leagths and the values of the angles and the displacements of the atoms from the equilibrium po-Elilons under vibrations and using the kinetic-energy cuefficients cal oil ared in the first part of the article (tay, yuppy seen, figite No. 4, 173, 1964), the author derives in general form relations for eni 1/3

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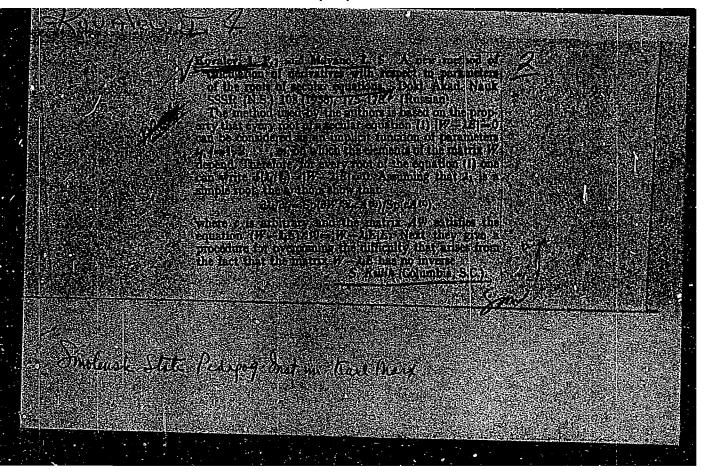
the calculation of the dispiscements of the atoms under normal vibrations in the XY, molecula. Formulas are also derived for the calculation of the electro-Optical parameters and the intensities of the infrared vibrational bands of monosilane and its deuterium substitutes. The displacements of the atoms, the electro-optical parameters, and the intensities are also calculated. Because of the high symmetry of the XY, molecula, the frequencies, shapes, and displaced ments of the stoms for yibrations of type F, do not depend on the interaction of the adjacent singles, but for vibrations of type E they depend on the interaction between the bonds themselves and between the angles and the bonds. The atom displacements in the vibrations of type A, do not depend explicitly on the masses of the atoms and the bond lengths, and in the case of E-type vibrations they are independent of the masses of the stoms. The electro-optical parameters are calculated in the first approximation of the valence—electro scheme on the basis of experimental data on the intensities (or \$184,) Comparison of the results ontained for \$184 with the data

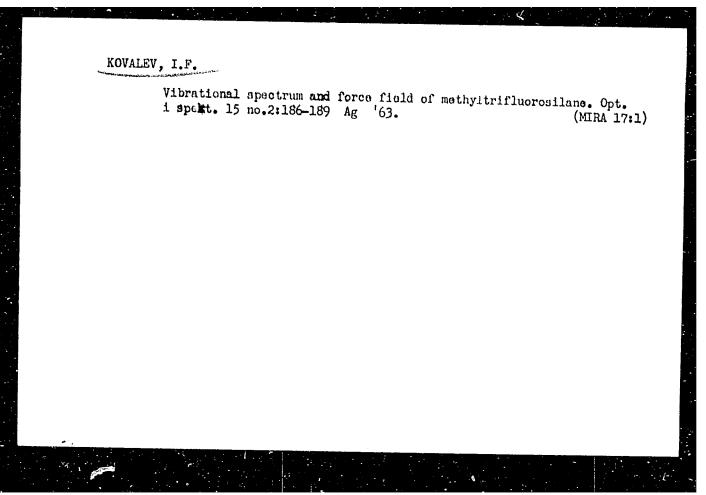
**5**513 - 373

VAGANOVA, I.P.; KOVALEV, I.F.

Calculation and interpretation of the vibrational spectrum of disilylmethene. Opt. i spektr. 17 no.6:960-961 D \*64.

(MIRA 18:3)





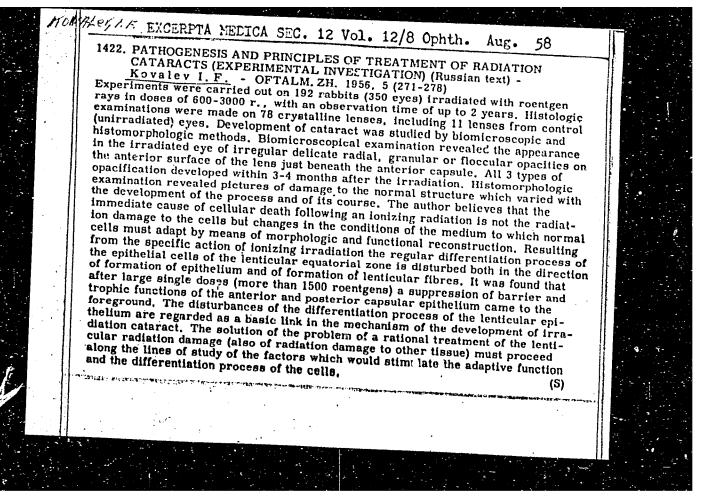
KOVALEV, I. F.

"Experimental Data on the Antagonism to Biological Action of Individual Portions of the Radiant Energy Spectrum." Cand Med Sci, Odessa State Medical Inst, Odessa, 1953. (RZhBiol, No 6, Nov 54)

Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (11)

SO: Sum. No. 521, 2 Jun 55

# Specificity of biological activity of ionizing radiations. Med.rad. 1 no.4:7-14 J1-Ag '56. (MIRA 9:12) 1. Iz Ukrainskogo nauchno-issledovatel'skogo eksperimental'nogo instituta glaznykh bolezney i tkanevoy terapii im. skad. V.P.Filatova (dir. - akad. V.P.Filatov) (RADIATIONS. eff. ionizing radiations on biol. substances)



KOVALEY, I.F.

The Specific Nature of the Biological Action of Ionizing Radiation," by I. F. Kovalev, Candidate of Medical Sciences, Ukrainian Experimental Institute of Ophthalmic Diseases imeni V. P. Filatov (director, Academician V. P. Filatov), Vestnik Rentgenologii i Radiologii, Vol 31, No 4, Jul/Aug 56, pp 54-63

Tests were run on infusoria (Paramaecium caudatum) in an effort to elucidate the specific nature of the biological effects of ionizing radiation. Radiation injuries caused by varying doses of X rays could be classified into three stages: 1) First stage in which infusoria were irradiated by small doses. A practically irreversible decreased capacity for adaptation to new conditions of external environment occurred, with evidence of some slowing of the functions of multiplication. (2) By using larger doses there was a decrease or loss of the function of multiplication and motion in addition to decreased adaptation, but the disturbance of conjugation and motion was easily reversible and was not transmitted to subsequent generations. (3) The third stage is characterized by the irreversible loss of the function of adaptation, reproduction, and motion and death under rays.

54M.1305

KOVALEVIIF

The author concludes that injuries appearing in the first stage and due to small doses are specific and are expressed in the decreased function of adaptation while all the other vital functions remain intact. The biological action of ionizing radiation in the second and third stages, however, are nonspecific and are expressed in a general denaturation of important enzyme and protein complexes.

The comparative relationship between intensity of ionizing radiation to specific and nonspecific injuries are: with the increase of dose there is an appropriate increase of intensity of nonspecific general-denaturation symptoms, but increasing the dose has very little effect on the specific

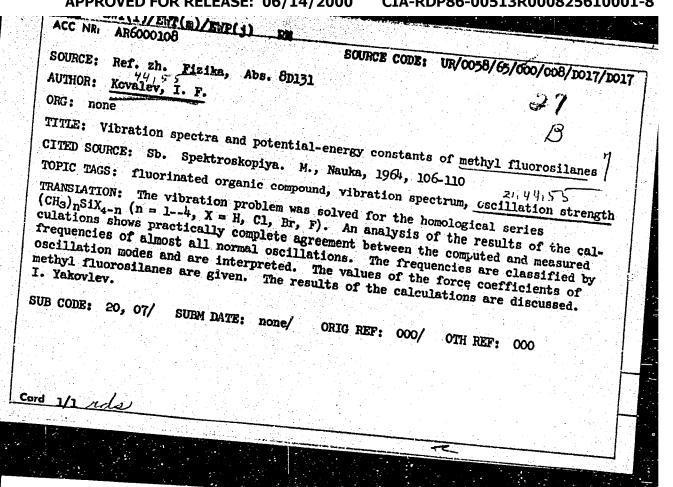
SUM. 1305

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Critique of certain basic concepts in radiobiology. Vest.rent.i rad.
33 no.3:52-57 Ny-Je '58

1. Iz Ukrainskogo nauchno-issledovatel'skogo akoperimental'nogo instituta glasnykh bolesney i tkanevoy terapii imeni akad. V.P. Filatova (RADIOLOY, radiobiol. basic concepts (Rus))

(BIOLOGY, basic concepts (Rus))
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KOVALEV, I.G., inzh.

Readjustable pneumatic-mechanical device for the assembling and tack welding of electric motor bodies. Svar.proizv. no.4:34-35 (MIRA 18:4)

1. Novosibirskiy turbogeneratornyy zavod im. XX s"yezda Kommunisticheskoy partii Sovetskogo Soyuza.

ROVALEV, I.F. (Leningrad)

Archival materials on hydraulic engineering and melioration.

Gidr. i mel. 16 no.9:60 S \*64.

(MIRA 17:11)

Calculation of the electro-optical parameters and intensities in the infrared spectra of molecules of the type XY<sub>4</sub>, XY<sub>3</sub>Z, and XY<sub>2</sub>Z<sub>2</sub>.

Izv. vys. ucheb. zav.; fiz. no.5:102-105 '64. (MIRA 17:11)

1. Saratovskiy pedagogicheskiy institut.

1 58896\_66 EPR/EYP(t)/EMP(b) Pe-4 1JP(t) 3D/ACH/JT ACCESSION RRI AP5019050 UR/0286/65/000/012/0077/0077 6C9.721.5 AUGIORI Kovalev ( C) Misheyevell, M.; Dolgov, V. V.; Bhpagin, B. V.;  $\mathcal{L}$ CIPLE: High-strength megnesium alloy, Class 40; No. 172050 BOURCE: Byulleten lzobreteniy Ltovarnykh znakov, no. 12, 1965, 77 TOPIC TAGE: magnesium alloy, high strength alloy, high strength magnesium alloy, ABSTRACT: This Author Ca-tificate introduces a high-strength magnesium alloy conteining zinc, cadmium, and zirronium: In order to improve mechanical properties and weldability, the alloy contains 2-15 zinc, 1-25 cadmium, 0.3-15 zirconium, 0.5-25 lauthanum, and the remainder is magnesium. ASSOCIATION: Organizatelya gosudaratvennogo komiteta po aviatelonnoy tekhnike SSSR (Organization of the State Committee on Aviation Engineering SSSR) EUBHITTED: 030ct63 No ref sov: 000 ENCL: 00 SUB CODE: SUB CODE: MA AS ATO PRESS: 4051 OTEER: 000 Cont 1/1(d)

PRUDKIN, Ya.M., gornyy inzh.; KOVALEV, I.G., gornyy inzh.; BATMANOV, Yu.K., gornyy inzh.

Effect of the increased advance rate or length of the longwall on the improvement of technical and economic indices.

Ugol' 37 no.1:22-24 Ja '62. (MiRA 15:2)

GRIDIN, A.D., inzh.; SAMSON, G.N., inzh.; PRUDKIN, Ya.M., inzh.; KOVALEV,

Ways of obtaining a record-high operative capacity of coal cutter loaders. Ugol' 37 nd.8:49-56 Ag '62. (MIRA 15:9)

1. Gosudarstvennyy proyektno-konstruktorskiy i eksperimental'nyy institut ugol'nogo mashinostroyeniya.

(Coal mining machinery)
(Coal mines and mining—Labor productivity)

KOVALEV, I.G., inzh.; KHEYLIK, V.Z., inzh.; SHADKHAN, V.M., inzh.

Efficiency in using new means of mechanizing ore mining operations at upper Kama potash mines. Gor.zhur. no.1:6-10 Ja 165.

1. Gosudarstvennyy proyektno-konstruktorskiy i eksperimental'nyy institut ugol'nogo mashinostroyeniya (for Kovalev, Kheylik).

2. Gosudarstvennyy soyuznyy institut po proyektirovaniyu predpriyatiy gcrnokhimicheskoy promyshlennosti (for Shadkhan).

KORALEY, T. S.

137-58-1-727

Translation from: Referativnyy zhurnal, Mekhanika, 1958, Nrl, p108(USSR)

AUTHOR:

Kovalev, I. G.

TITLE:

A Die for Cutting Hexagonal M8, M10 and M12 Bolt Heads (Shtamp dlya vyrubki shestigrannykh golovok boltov M8, M10

PERIODICAL: Mashinostroitel', 1937, Nr 5, p 38

ABSTRACT:

A die for cutting hexagonal heads for M8, M10 and M12 bolts has been developed and introduced. The blank used is a turned or cold-upset bar with a round head cut on six sides in dies of EI-161 or 9KhS steel. The clearance between the die halves when the plunger is in the low position is 0.3-0.5mm. The die is provided with a guard and guides.

1. Dies-Development 2. Dies-Characteristics

V. Ya.

Card 1/1

KOVALEV I.C.

Translation from: Referativnyy Zhurnal, Mashinostroyeniye, 1957, 123-1-450

Nr 1, p. 75, (USSR)

AUTHOR:

Kovalev, I. G.

TITLE:

Trimming and Blanking Die for Holes for Threading (Shtamp dlya otbortovki otverstiya pod rez'bu i vyrubki kontura)

PERIODICAL: Sbornik ratsion.predlozh. Min-vo Elektrotekhn.Prom-sti SSSR, 1956, Nr 3 (61), p.17-18

ABSTRACT:

Trimming holes in workpieces was done before in a second operation after the part was blanked, and often creating the spoilage along the draws. A new, combined die with special trimming punch is proposed. This one will simultaneously pierce the hole and trim it with the progressive movement of the slide; then, after the stock is moved two stops, the part is blanked. To evoid ragged edges it is recommended to use the die in a slow-operating press.

Card 1/1

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KOV BLEY, IT, G.

137-58-1-1884

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 1, p 254 (USSR)

AUTHORS: Kovalev, I.G., Kovaleva, Z.N.

TITLE: Hardening of Mg-Zn-Zr Alloys by a Combination of Heat and Mechanical Treatment (Termomekhanicheskoye uprochneniye

splavov sistemy Mg-Zn-Zr)

PERIODICAL: V sb.: Metallurg. osnovy lit'ya legkikh splavov. Moscow,

Oborongiz, 1957, pp 429-441

ABSTRACT: The properties of alloys of the Mg-Zn-Zr system, VM65-3

and VM65-1, are described. Alloys VM65-1 and VM65-3 are strengthened by aging, in the process of which a finely dispersed hardening phase MgZn2 is liberated. Various aging procedures are examined. The following optimum aging regime after hot extrusion is recommended: temperature 1500, holding 24 hours. Annealing on extruded semi-finished products at 3500 for 1 hour leads to softening as a result of re-crystallization and partial solution of the hardening phase, and therefore annealing should not be performed if high mechanical properties are to be preserved. Re-extrusion of alloys provides a finer crystalline

Card 1/2 structure and increases strength as compared to a single

137-58-1-1884

Hardening of Mg-Zn-Zr Ailoys by a Combination of Heat (cont.)

extrusion. Hot drawing at  $240\text{-}250^{\circ}$  reveals a 3-5 percent increase in the strength of extruded blanks.

P.V.

1. Alloys-Hardening 2. Alloys-Heat treatment

Card 2/2

KOVALEV, I. G.

Davydov, Yu. P.; I. G. Kovalev; and G. V. Pokrovkiy. Special Features of Sheet Forming of Aircraft Steel and Aircraft Alloys. p.103

Pressure Treatment of Alloys; Collection of Articles, Moscow, Oborongiz, 1958, 141pp.

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The second of th	The second secon	:	Membres, S.Ms. Effect of Bare-Earth Metals on Sulfur Distribution and Sulfur Fistribution and	PART V. RASE METALS IN STREETS	School Related Properties of Marchitage Attention Alloys of the Magnesia-Aury Bessel Magnesia-Aurymana-Certa Systems  1052 and Magnesia-Aurymane-Certan Systems  1052 and Magnesia-Aurymane-Certan Systems	Mitherer 7 M and 7 To Magnesian Alloys With Rare Metals		HET Research Meals	Vrought Mag	lary Metals (Cont.)	: :	ALLOIS WITH BLEE-PETAL LOSITIONS	mentonductive alloys) and discuss. No personalities are sentioned. So many the control of the articles.	examplyings for antogobile electrical is a worker, and material reliable for fact of the addition of castain elegation on the properties of the control of the state of castain elegation on the properties of the booking the	on properties of sagmestim alloys and stools is analyzed. The mass of the	ditions of rare metals are presented and discussed along with treating, renading, miching, and their alloys. The effect of which they sight	exasted at the First 111-180 contents to the party which were presented and dis- stitute of Metalling J. Ledgey of Sciences on Accessed 1110ys, build in the II. In the content of the con	The may also be used by students of schools of higher education industries.  COTTAINS: The collector.	FUPCAR: This collection of articles is intended for metallurgical and	Ed.: I.E. Shapovalov; Ed. of Publishing House: O.H. Kanayers; Tech. Ed.; P.G. Islant'yers.	Sponsoring Agreedes: Akadesiya nauk 8888. Inatitut metallurdij. USun Kontesiya po redkim metallan pri nauchno-takkatokankun kontes	First All-Outen Conference on Expendent Alleys) Moscow, Metallurgistat, 1900, 438 p. 3,130 copies printed.	Redrige socially 1 splays, touts in the redrike metallow. Let, Mascow, 1957	TRACE I BOOK EXPLOITATION SOT/4164		
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8/136/60/000/05/012/025 E071/E235

AUTHORS: Kazakov, A. A., /Kovalev, I. G., and Kolpashnikov, TITLE:

Heat Resistant Deformable Magnesium Alloy MA13

PERIODICAL: Tsvetnyye metally, 1960, Nr 5, pp 62-65 (USSR)

ABSTRACT: On the basis of preliminary investigations of various magnesium alloys, carried out during 1956 to 1957 by VIAM, and literature data, an alloy of the system Mg-Th-Mn under the name of MA13 (similar in composition to an American alloy NM21KhA) was found to be the most heat resistant and was chosen for more detailed investigations; the results of these are reported in the paper. A few heats of the alloy were prepared for the investigation in a steel crucible (12 kg) with the application of flux VI2. Magnesium and alloying addition MGS-1 was melted at 700 to 720°C. Thorium was introduced in the form of turnings at 800°C in a preheated bell. During the introduction of thorium, the surface of the metal bath was covered with a small amount of flux containing 55% of KCl, 28% of CaCl<sub>2</sub>, 15% of BaCl<sub>2</sub> and 2% of CaF<sub>2</sub>. The alloy (cooled to about 720 to 740°C) was cast into metal moulds, preheated to 100 to 150°C. The experimental ingots (25 x 150 x 300 mm) were rolled into sheets 1 to 6 mm

69831 S/136/60/000/05/012/025 E071/E235

Heat Resistant Deformable Magnesium Alloy MA13

thick, on a two high mill, with rolls 4000 mm in diameter, preheated to 100 to 120°C. Temperature at the beginning of rolling 450 to 500°C, at the end of rolling 300 to 350°C, reduction per pass 20 to 30%. Rolled sheets were thermally treated with an intermediate cold rolling:
a) heating (for hardening) to 550 to 560°C with a 30 minute soaking in a protective atmosphere (sulphurous gas) and cooling in air; b) cold rolling with total reduction of 7 to 10%; 'c) ageing at 200°C for 16 hours. After hardening, the sheets were pickled in a 5% solution of nitric acid and hand dressed. After hot rolling, the alloy possessed a fibrous structure of a deformed, partially recrystallised material. After hardening, a fully recrystallised equiexial structure is formed. The physical properties of the alloy are entered in Table 1; the mechanical properties are given in Table 2; comparison of the mechanical properties of the alloys MA11, MA2-1, MA8 with those of MA13 are given in Tables 3, 4 and Fig 4. It was found that at temperatures above Card 2/3 240°C alloy MA13 possesses superior mechanical properties

DAVIDOVA, G.V.; LOYTER, M.N.; ALEKSEYEV, N.A.; KOVALEV, I.I.; DANILOV, A.Ye.;
SHENDRIKOV, G.L., 1.0. glavnogo metodista; ORLOVA, V.P., redaktor;
PAVIOVA, M.M., tekhnicheskiy redektor

["Water resources management and rural hydroelectric power stations"
pavilion; a guidebook] Pavil'on "Vodnoe khoziaistvo i sel'skie
gidroelektrostantsii"; putevoditel'. Moskva, Gos. izd-vo selkhoz.
lit-ry, 1956. 21 p. (MIRA 9:12)

1. Moscow. Vsesoyuznaya sel'skokhozyaystvennaya vystavka, 19542. Direktor pavil'ona (for Danilova)

(Moscow--Agricultural exhibitions)

(Water supply, Rural)

(Hydroelectric power stations)

KOVALEV, I.I., inzh. leunogo khozyayatva (stantsiya Sukhinichi)

New method of shelterbelt tillage. Put'i put. khoz. no.4:33-34
Ap '58.

(Windbreaks, shelterbelts, etc.) (Plows)

Films of gallium arsenide and their properties. V. A. Presnov, L. G. Lavrent'yeva, M. D. Vilisova, <u>I. K. Kovalev.</u>

On the physico-chemical nature of the formation of contacts of gallium arsenide with metals. V. A. Presnov, A. N. Vyatkin. (Presented by A. N. Vyatkin--10 minutes).

Report presented at the 3rd National Conference on Semiconductor Compounds, Kishinev, 16-21 Sept 1963

ACC NR. AR6000072 SOURCE CODE: UR/0275/65/000/009/B011/B011

AUTHOR: Lavrent'yeva, L. G.; Kovalev, I. K.

27

TITLE: Preparation of single-crystal germanium film by sputtering in vacuum

SOURCE: Ref. zh. Elektronika i yeye primeneniye, Abs. 9889

REF SOURCE: Dokl. Nauchno-tekhn. konferentsii, posvyashch. dnyu radio. Tomsk, Tomskiy un-t, 1964, 3-6

TOPIC TAGS: single crystal, semiconductor single crystal, germanium single crystal

ABSTRACT: Epitaxial films of Ge were obtained at substrate temperatures of the order of  $450-500^{\circ}\mathrm{C}$ . Ge of p-type was sputtered from a tungsten spiral onto an n-type Ge substrate. The source temperature was 1100 to  $1300^{\circ}\mathrm{C}$ . The base temperature was maintained by an external heater within a range of 450 to  $650^{\circ}\mathrm{C}$ . The pressure in the system was  $10^{-5}$  mm Hg. The optimum sputtering rate was found to be  $1\,\mu/\mathrm{min}$ . At high sputtering rates a film with a fine-grained deposit on the surface was obtained. The film thickness was 3 to  $15\,\mu$ . Sputtering of a p-type film on an n-type Ge substrate resulted in the formation of p-n junctions. The contact was fused into the film with pieces of Sn in air at a temperature close to the melting point of Sn. Films not protected by Sn were etched away with perhydrol.

**Card 1/2** 

fication coefficient of 260 200 Mamp. The relatively he concentration defects in th	tics of the obtained p-n junction of 520 Has digh back currents were apparentle film. Refs.: 3.	ons indicated a recti- pp, and a back current of ly caused by the higher N.Sh.
BUB CCDZ: 20/, 👙 SUBM DAT	E: none	n.on.
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ZHDANOVSKIY, N.S.; KOVALEV, I.M.; KHASHCHINSKIY, V.P., professor.

[Rural thermal electric power stations] Sel'skie teplovye elektrostantsii. Pod red. V.P.Khashchinskogo. Moskva, Gos. izd-vo sel'khoz. lit-ry. 1953. 123 p. (V pomoshch' sel'skim elektrifikatorem)

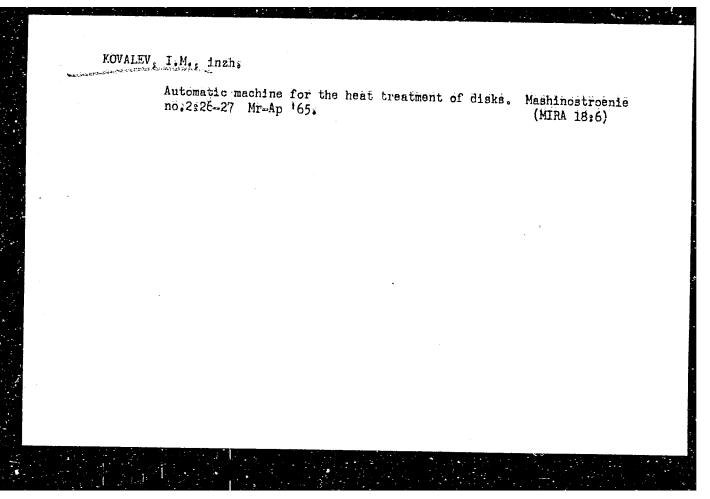
(MIRA 7:3)

(Electric power plants) (Heat engines)

KOVALEV, I.M., inzh.; KOZHUKHARI, G.1., inzh.

Annealing malleable cast iron in continuous electric furnacec with a protective atmosphere. Mashinestrochie no.1:70 Jauf 165.

(MIRA 18:4)



L 3272-66 EWT(m)/EWP(v)/T/EWP(t)/EWP(k)/EWP(b)/EWA(c) JD/HM

ACC NRI AP5025606

UR/0135/65/000/010/0004/0006

621.791.75.01:538.122

AUTHOR: Kovalev, I. M. (Engineer)

TITLE: Deflection of welding arc in a trensverse magnetic field

SOURCE: Svarochnoye proizvodstvo, no. 10, 1965, 4-6

TOPIC TAGS: arc welding, transverse magnetic field, plasma jet, welding electrode,

tungsten

ABSTRACT: The deflection of the welding arc in a transverse magnetic field of constant intensity is the most elementary form of the interaction between the arc's magnetic field and the controlling magnetic field. The pattern of this deflection is determined by an analysis of the forces acting on the arc-discharge plasma. In this connection the authors present the results of an experimental investigation performed to verify the theory that the model of a flexible conductor with current applies to this arc. A homogeneous magnetic field was created by means of coils. The relationship between the intensity of the transverse magnetic field I, the arc length I, and the current intensity I was determined by means of a device consisting of electrode 1 and fixed anodes 2 and 3 isolated from one another and connected in parallel to the circuit of the arc current (see Fig. 1 of the Enclosure), on fixing at e.g. 60° the angle of deflection of a DC direct-polarity (W-Cu) arc burning in

Cord 1/3

L 3272-66

ACC NR: AP5025606

argon (tungsten electrode) within the transverse magnetic field. On this basis, it is shown that the compression of the arc column by the intrinsic magnetic field and its stabilization by the plasma jet endow, the arc with the properties of a flexible conductor with current, whose deviation in the transverse magnetic field obeys Ampere's law. Thus, the specific property of the arc as a plasma conductor consists in that the Lorentz forces induced in the arc column by the intrinsic magnetic field of the moving charges are directed toward the column's center. These forces, compressing the column, create a pressure field whose gradient compensates the Lorentz forces:

$$\operatorname{grad} p = \frac{1}{c} \left[ \overline{J} \, \overline{H} \right]. \tag{1}$$

On this basis, the author derives empirical formulas for determining the deflection angle and anode-spot displacement of the arc in the presence of a transverse magnetic field of constant intensity. Orig. art. has: 7 figures, 11 formulas.

ASSOCIATION: MVTU im. Baumana 1415

SUBMITTED: 00

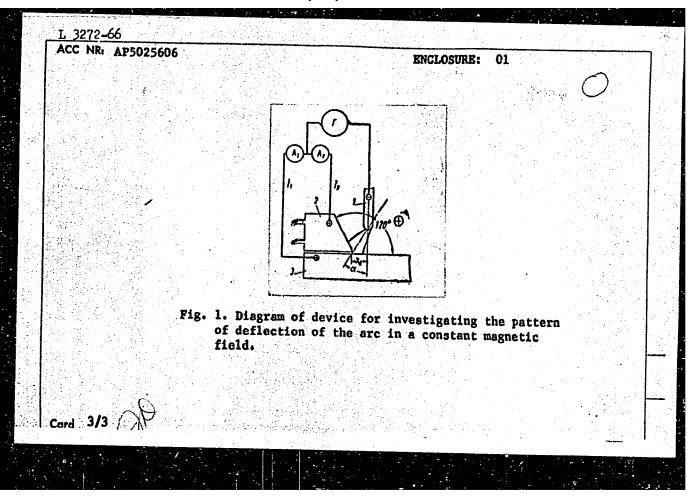
ENCL: 01

SUB CODE: IE, EM

NO REF SOV: 004

OTHER: 001

Card 2/3



L 3271-66 EVT(m)/EWP(v)/T/EWP(t)/EWP(k)/EWP(b)/EWA(c) JD/HM

ACC NRI AP5025607

UR/0135/65/000/010/006/0009 621.791.75.01:538.122

AUTHOR: Kovalev, I. M. (Engineer); Akulov, A. I. (Doctor of technical sciences)

TITLE: Stability of welding arc in a transverse magnetic field

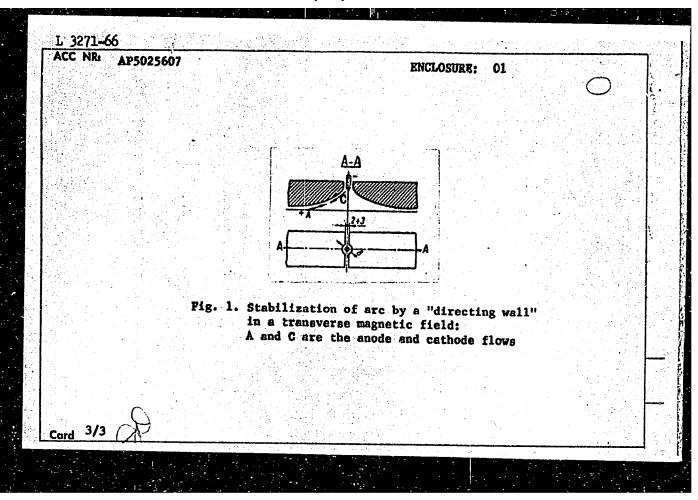
SOURCE: Svarochnoye proizvodstvo, no. 10, 1965, 6-9

TOPIC TAGS: arc welding, transverse magnetic field, welding arc stability, welding electrode

ABSTRACT: The possibility of controlling the stability of the welding arc by means of a transverse magnetic field is markedly restricted by the arc's inability to elongate without disrupting when deflected by such a field. It is shown that this restriction may be to a large extent eliminated by stabilizing the arc by means of the flow of the protective gas. This results in equalizing the plasma velocity field and increasing the total velocity of the cathode flow and thus in deforming the arc as a single whole, particularly in the upper (near-cathode) region of the column. Another method of stabilizing the arc is the employment of the so-called "directing wall". In this case, the arc discharge moving in a commercial-frequency transverse magnetic field is bounded by two graphite or cooled copper blocks (see Fig. 1 of the Enclosure). A tungsten electrode and the protective gas are passed through the space between the blocks. The flow rate of the argon is 7-8 liters/min, which precludes

Card 1/3

ACC NR: AP502560	<b>7</b>			3 1			
compression of the arc by the gas flow. The directional profile of the block is parabolic, thereby maximizing the displacement of the anode spot. The device stabilizes the flame so that the anode flow bounces from the walls and is directed counter to the cathode flow, thus displacing the anode spot. Orig. art. has: 11 figures.							
ASSOCIATION: MVT							
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KOVALEV, I.M., inzh.

Deflection of welding arcs in a transverse magnetic field. Svar. proizv. no.10:4-6 .0 '65. (MIRA 18:10)

1. Moskovskoye vyssheye tekhnicheskoye uchilishche im. Baumana.

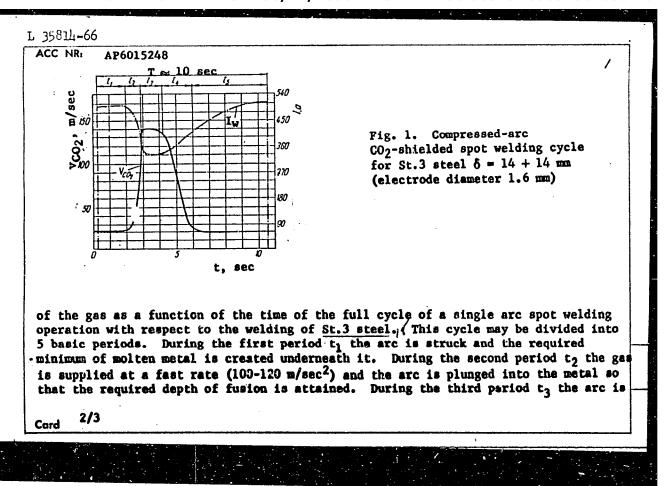
KOVALEV, I.M., Inch.; AKOLOV, A.I., doktor tekhn. nauk

Atvidility of a welding arc in a transverse magnetic field.

Genr. proize. no.10:6-9 0 '65. (MIRA 18:10)

1. Moskovskoye vyssheye tekhnicheskoye uchilishche im. Baumana.

ACC NRI	AP6015248	(A) soi	URCE CODE: UR/	0125/66/000/005/0056/0057 33
AUTHOR:	Kovalev, I. M.	( ' ' )		32
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	stov-on-Don Ins sel'khozmashin		tural Machine Bu	ilding (Rostovskiy-na-Donu
TITLE:	Compressed-arc	spot welding of jo	oints	
SOURCE:	Avtomaticheska	ya svarka, nc. 5,	1966, 56-57	steel,
TOPIC TA steel	GS: arc spot w	elding, arc weldin	ng, spot welding	g, welding technology/St.3
jet of C fusion; owing to ment in neath th the weld leading since th creases	O2 during gas-e the latter inc both the atten the conditions e arc. Hence, ling rate is zer to a rapid incr e overall amoun proportionately	electric welding materials and the melting power to, i.e. during arcease in the extendat of the molten electric than time. This, let	the decreasing the linear energiates the linear energiates of the compress spot welding. It of the molten lectrode metal anowever, can be	that the use of a high-speed to increase the depth of the arc, by of the arc and the improvent and the molten layer undersed arc may be maximized when This has the disadvantage of layer underneath the arc, and molten base metal incremedied by adjusting the 1 presents the flow rate vCO2
Card 1	12			UDC: 621.791.8



neld fast and luced to its o	15248 the spot base is v	the arc beg	ins to weld up	the hole. Du	ring the fifth
$steel \delta = 14. +$	welding of the spo 14 mm): electrod electrode reach 10	e diameter	1.6 mm, voc. 1	30-150 m/sec.	I - 480-520 a
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KOVALEV, I.N. [deceased]; TRYAPITSYNA, L.N.

Bibliographic index of works completed at the station of the Astrakhan Preserve. Trudy Astr. 2ap. no.5:353-369 '61.

(Bibliography--Astrakhan Preserve--Natural history)

(Astrakhan Preserve--Natural History--Bibliography)

KOVALEV, I.P.; PROKOPENKO, A.P.; TITOV, Ye.V.

Spectroscopic study of some unsaturated six-membered lactones. Ukr. khim. zhur. 29 no.7:740-743 '63. (MIRA 16:8)

1. Khar'kovskiy nauchno-isslovatel'skiy khimiko-farmatsevticheskiy institut.

(Lactones-Spectra)

# KOVALEV, I.P.; TITOV, Ye.V. Infrared absorption spectra of natural compounds. Part 1: Flavonoids. Absorption bands of carbonyl and hydroxyl groups. Zhur. ob. khim. 33 no.5:1670-1676 My '63. (MIRA 16:6) 1. Khar'kovskiy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut. (Flavonoids) (Carbonyl group—Absorption spectra) (Hydroxyl group—Absorption spectra)

DAL!, V.I.; ZMIYEVSKIY, P.K.; KOVALEV, I.P.

Heavy refining residues of Volgograd potroleums as raw materials for the retarded coking process. Izv. vys. ucheb. zav.; neft' i gaz 6 no.10:55-58 '63. (MIRA 17:3)

l. Dne propetrovskiy khimiko-tekhnologicheskiy institut im. Dzerzhinskogo.

L 33271-66

ACC NR: AR6016193

SOURCE CODE: UR/0058/65/000/011/D025/D025

AUTHOR: Kovalev, I. P.; Titov, Ye. V.

TITIE: Infrared absorption spectra of natural derivatives of α and γ pyrone

SOURCE: Ref. zh. Fizika, Abs. 110191

REF SOURCE: Tr. Komis. po spektroskopii. AN SSSR, t. 3, vyp. 1, 1964, 637-643

TOPIC TAGS: ir spectrum, absorption band, spectrum analysis, hydrogen bonding, chelate compound

ABSTRACT: The authors investigated the ir absorption spectra of 55 flavoncids, coumarins, and furocoumarins, and propose a classification of the frequencies in the  $4000-650~{\rm cm^{-1}}$ . It is shown that spectroscopic identification of the derivatives of  $\alpha$  and  $\gamma$  pyrone by means of the frequencies and intensities of the absorption bands of the hydroxyl, carbonyl, and other groups is possible. The chelate hydrogen bond of the flavonoids, which plays an important role in the manifestation of their biological action, is investigated. Work is done on the preparation of a chart of ir spectra of the derivatives of  $\alpha$  and  $\gamma$  pyrone (standardization of the measurement of the spectrum, development of type of documentation). [Translation of abstract]

SUB CODE: 20, 07

Card 1/1 Py

KOVALEV, 1.P.; LITVINENKO, V.I.

Flavenoid glycosides. Part 1: Monoglycosides. Khim.prirod. soed. no.4:233-241 '65. (MIRA 19:1)

1. Khar'kovskiy nauchno-issledovatel'skiy khimiko-farmatsevti-cheskiy institut. Submitted March 23, 1965.

KOVALEV, I.P.; TITOV, Ye.V.; CHERNOBAY, V.T.; KOMICCARENKO, N.F.

Infrared spectra of glucosides of the strophanthidin series.
Ukr.khim.zhur. 31 no.5:513-516 '65.

(MIRA 18:12)

1. Khar'kovskiy nauchno-issledovatel'skiy khimiko-farmatsevticheskiy institut. Submitted Dec. 6, 1963.

KOVALEV, I.P., starshiy inzh.

Prerefining of petroleum. Neftianik 6 no.4:15-17 Ap '61.

(MIRA 14:8)

1. Stalingradskiy sektor Spetsial'nogo konstruktorskogo
byuro po avtomatike v neftepererabotke i neftekhimii.

(Stalingrad--Petroleum--Refining)

KOVALEV, I.F., inzh.

Redesigning a pressure-vacuum distillation unit. Neftianik 6 no.9:16-17 5 '63. (LIE 14:10)

1. Stalingradshiy neftepererabatyvayushchiy zavod. (Distillation apparatus)

Coursely

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: Form Animals, The Honoglee.

APPROVED FOR RELEASE: 06/14/2000 | NCJA-RDR86-00513R000825610001-8"

Author Institut. : Movalev, I. S.

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: My Experience in Substituting Queens.

Orig Pub.

: Pehelovodstvo, 1958, No 6, 20-22

abstract

: Into a colony with a 2-years old queen a mature queen cell was placed at the time of intersive gataering [of honey] into the second entrance of the upper bedy of the hive or into the storage part, without removing the old queen. in 20 colonies out of 24 the change of the queenc proceeded satisfactorily. It was observed that the queen accepts more readily strong colonies whose development has been completed. If the colony does not accept the

Card:

1/2

Uard:

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VISHNEVSKIY, Nikolay Yevgen'yevich; GIUKHANOV, Nikolay Parmenovich; KOVALEV, Ivan Sidorovich; STOLYAROV, V.I., retsenzent; MERKIN, G.I., kandidat tekhnicheskikh nauk, redaktor; CHERHOUSOV, N.P., inzhener, redaktor; GOFMAN, Ye.K., redaktor izdatel'stva; SOKOLOVA, L.V., tekhnicheskiy redaktor

[High pressure apparatus with hermetically sealed electric motors]
Apparatura vysokogo davleniia s ekranirovannym elektrodvigatelem.
Moskva, Gos. nauchno-tekhn. izd-vo mashinostroit. lit-ry, 1956.
178 p. (MIRA 9:8)

(Electric motors) (Machinery industry)

KOVALEVITS.

Category : USSR / Radio Physics. Radiation of Radio Waves. Trans-I-5

mission Lines and Antennas

Abs Jour : Ref Zhur - Fizika No 3, 1957, No 7287

Author : Kovalev, I.S.

Title : Calculation of Wave Attenuation in Roctangular Waveguides With the Aid of the Complex Electromagnetic Field Vector.

Orig Pub : Tr. Ryazansk. radiotekhn. in-ta, 1956, 1, 51-58.

Abstract : The calculations are carried out by directly solving Maxwell's equations with the aid of the complex electrimametic field vector. In conclusion the author indicates certain inaccuracies that other investigators have overlooked in the derivation of a formula for the attenuation of the H<sub>10</sub> wave in rectangular waveguides.

Card : 1/1

- 29 -

### CIA-RDP86-00513R000825610001-8" APPROVED FOR RELEASE: 06/14/2000 5/142/62/005/002/011/019 E200/E382

9.1400

AUTHOR: Kovalev, I.S.

Theory of the asymmetric air-filled strip TITLE: transmission line

Izvestiya vysshikh uchebnykh zavedeniy, Radiotekhnika, v. 5, no. 2, 1962, 245 - 249

The theory of an asymmetric air-filled strip transmission line is outlined. Taking the boundary fields into consideration formulae are derived for the transmitted power P, power loss per unit of the line Q, and the attenuation  $\beta$ . It is found by treating the strip line as an ideal condenser and using conformal mapping methods that:

$$P = \frac{C}{8\pi} \cdot \frac{E_o^2}{Z} \cdot d^2(r_B - r_A)$$
 (11)

where E is the homogeneous field in the cavity,

Card 1/4

5/142/62/005/002/011/019 E200/E382

Theory of ....

 $Z = \sqrt{\mu/\epsilon}$  is the wave impedance of free space, b is the width of the strip,

d is the distance between the strip and the conducting plane,

 $\mathbf{r}_{\mathrm{A}}$  and  $\mathbf{r}_{\mathrm{B}}$  are the roots of the transcendental equation:

$$r - \ln r - 1 - \frac{b\pi}{2d} = 0$$
 (10).

It is further found that:

$$Q = \frac{C}{4\pi^2} \cdot \frac{R_n}{Z^2} \cdot E_0^2 \cdot d \cdot \left[ (r_B - r_A) + \ln \frac{e^{r_A} + 1}{r_B} \right]$$
 (17)

where  $R_n = \sqrt{\omega\mu/8\pi\gamma}$  is the surface impedance;

Card 2/4

CIA-RDP86-00513R000825610001-8" APPROVED FOR RELEASE: 06/14/2000

Theory of ....

S/142/62/005/002/011/019 E200/E382

$$\beta = \frac{R}{z} \cdot \frac{1}{d} \left[ 1 + \frac{\ln \frac{e^{A} + 1}{e^{B} + 1}}{r_{B} - r_{A}} \right]$$
 (18)

These formulae are represented by Fig. 4 (P/E<sup>2</sup> as a function of b/d), Fig. 5 (Q/E<sup>2</sup> as a function of b/d) and Fig. 6 ( $\beta$  as a function of b/d). A table of roots of Eq. (10) for the permissible range of b/d ratios is given. Experimental verification of the power-transmission formula has demonstrated an agreement within 7% of the theoretically computed value. There are 6 figures and 1 table.

Card 3/4



APPROVED FOR RELEASE: 06/14/2000

CIA-RDP86-00513R000825610001-8

# KOVALEV, I.S.

Calculation of the capacitance of the characteristic impedance of a nonsymetrical strip line. Izv. vys. ucheb. zav.; radiotekh. 5 no.3:368-375 My-Je '62. (MIRA 15:9)

1. Rekomendovana kafedroy teo eticheskikh osnov elektrotekhniki Ryazanskogo radiotekhnicheskogo instituta. (Condensers (Electricity))

414312

S/142/62/005/004/009/010 E192/E382

9.1400

AUTHOR:

Kovalev, I.S.

TITLE:

Determination of the capacitance and the characteristic impedance of a symmetrical strip line with an air-

filler

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy, Radiotekhnika, v. 5, no. 4, 1962, 527 - 530

TEXT: An exact analytical determination of the capacitance of a strip line presents considerable difficulties since it depends on the solution of elliptical integrals of the first order. The capacitance of such a symmotrical line was therefore determined experimentally, the measurements being conducted at a frequency of 60 Mc/s by means of a Q-meter. It is found from the experimental data that the capacitance can be approximately described by the following formula:

$$C = 0.15 \left(1 + \frac{b}{d}\right) \frac{\eta F}{cm}$$
 (2)

where b is the width of the central strip of the line and Card 1/3

S/142/62/005/004/009/010 E192/E382

Determination of the ....

d is the spacing between the strip and the earthing plates. The formula is valid for b/d > 0.6. The impedance of the strip can be expressed by:

$$Z_{o} = 30/C$$
 (1).

If  $Z_0$  is evaluated by using Eq. (2) for expressing C, the result is fairly accurate, except when the thickness  $\Delta$  of the centre strip is a significant fraction of d. In the latter case, the capacitance can be expressed by:

$$C=0.15\left(1+\frac{b}{d}\right)\left(\frac{1}{1-\frac{\Delta}{d}}\right)\frac{nF}{cm} \qquad for \quad \frac{b}{d}<2$$

$$C = 0.15 \left[ 1 + \frac{b}{d} \left( \frac{1}{1 - \frac{b}{d}} \right) \right] \frac{nF}{cm} \qquad \text{for } \frac{b}{d} > 2$$

Card 2/3

"Card 3/3